





PREPARATION AND CHARACTERIZATION OF FUNCTIONAL METALLIC NANOPARTICLES AS POTENTIAL DRUG DELIVERY SYSTEMS

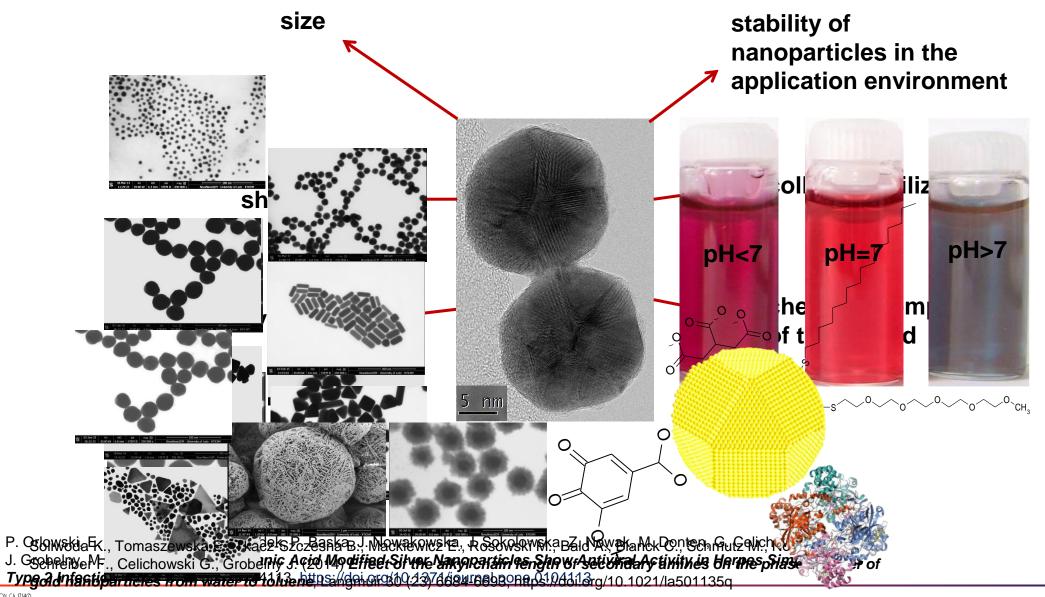
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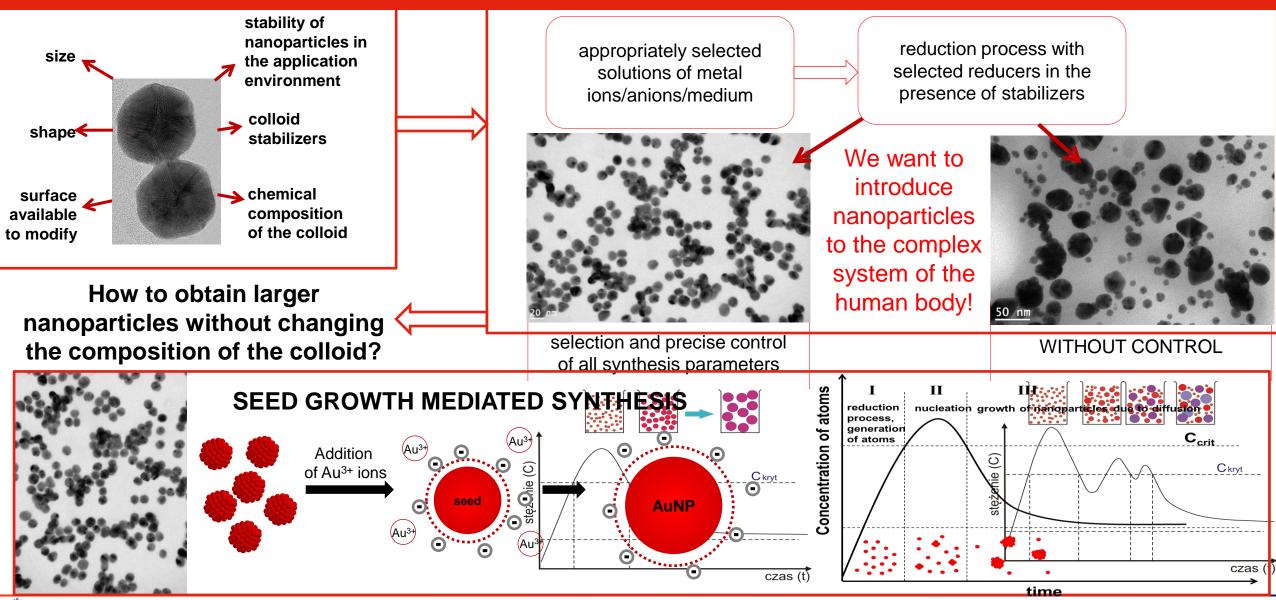
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What determines the biological activity of nanoparticles?

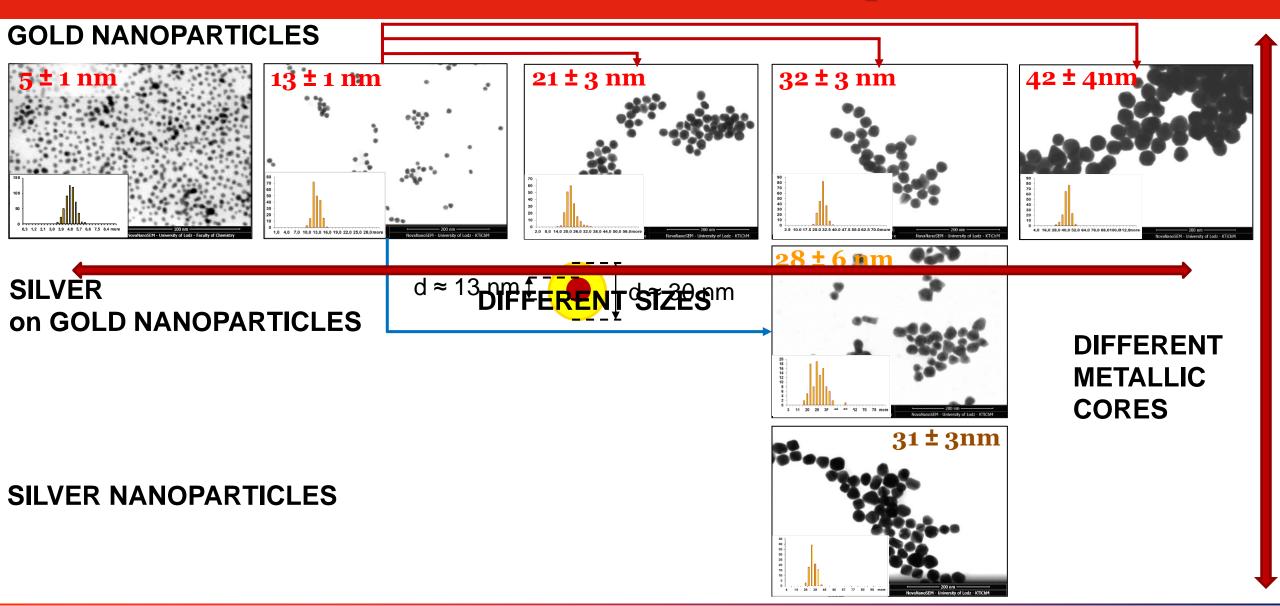


Synthesis of metallic nanoparticles by chemical reduction method



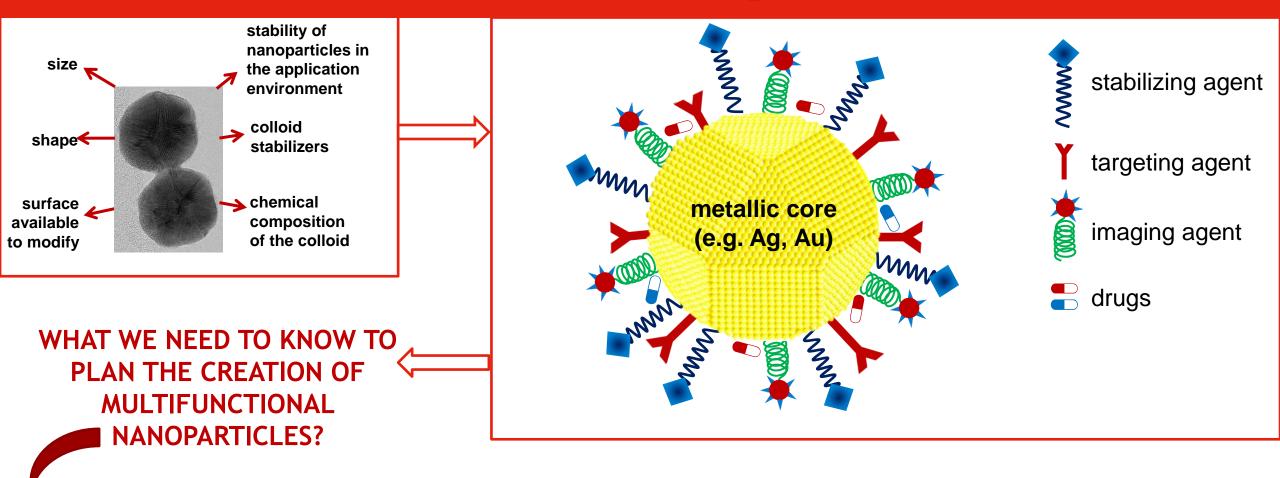
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Obtained colloids of metallic nanoparticles





Multifuncional nanoparticles



How many particles of different types of modifier have been absorbed on the metal surface?

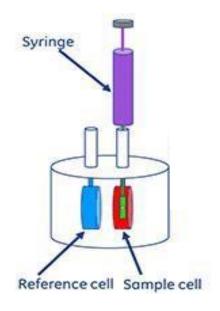
or

How many modifier molecules are needed to obtain full coverage of the nanoparticle surface?

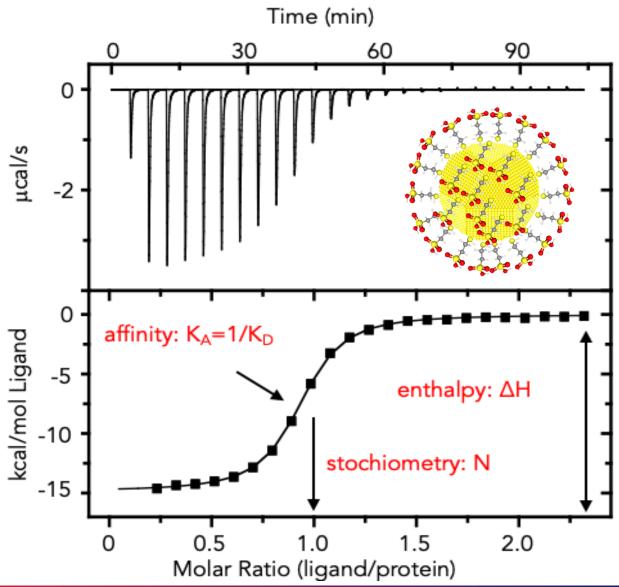


How to determine the amount of particles adsorbed on the nanoparticles surface?

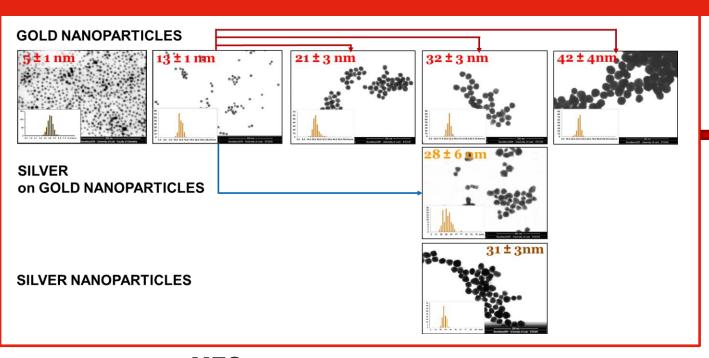
Isothermal titration calorimetry (ITC) is a technique used to quantitatively study interactions between molecules. It directly measures the heat that is released or absorbed during bond formation. The measurement method allows to determine the binding affinity, stoichiometry, entropy and enthalpy of the bond formed between the reactants in the solution (e.g., binding stoichiometry, ΔH , ΔS , and ΔG).







Materials determined using ITC



Precise knowledge of the metal mass in the colloid and the nanoparticles size allows to determine the molar concentration of nanoparticles and to estimate the surface area available for modification.

 $HS-PEG-SO_3Na$ M = 422,58 g/mol

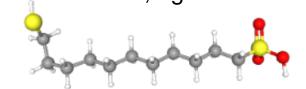
MES

Sodium methanethiolate M = 164,18 g/mol



MUS

11-Mercaptoundecane-1-sulfonic acid sodium salt M = 290,4 g/mol

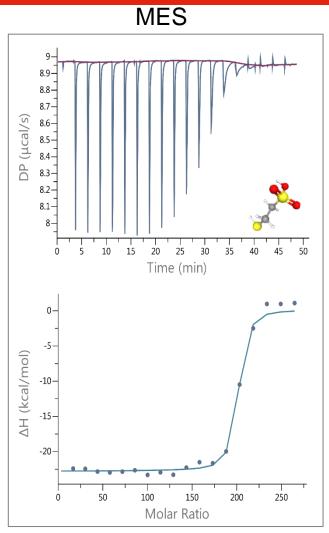


DIFFERENT LENGTH OF THE CARBON CHAIN

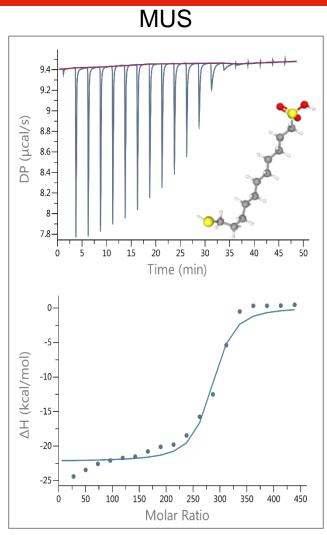


AuNPs_5 nm

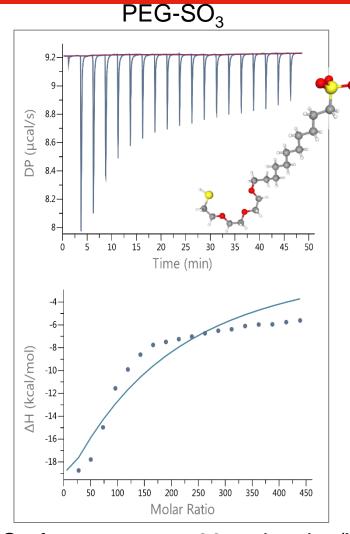




Surface coverage 203 molecules/NPs, that is 2,5 molecules/1 nm²



Surface coverage 291 molecules/NPs, that is 3,5 molecules/1 nm²



Surface coverage 86 molecules/NPs, that is 1,25 molecules/1 nm²



Obtained results for different sizes of nanoparticels

CALCULATED DEGREE OF COVERAGE [molecule/1nm²]

	AuNPs_5nm	AuNPs_13nm	AuNPs_20nm	AuNPs_30nm	AuNPs_40nm
MES	2,5	4	5	4,5	6,3
MUS	3,5	5,25	6	5,5	8
PEG-SO ₃	1,25	**	**	**	**

INCREASE IN DIAMETER

INCREASE IN THE NUMBER OF PARTICLES PER 1nm²

** - at least two types of binding of molecules to the metal surface, results under analysis



Obtained results for different mettalic cores of nanoparticels

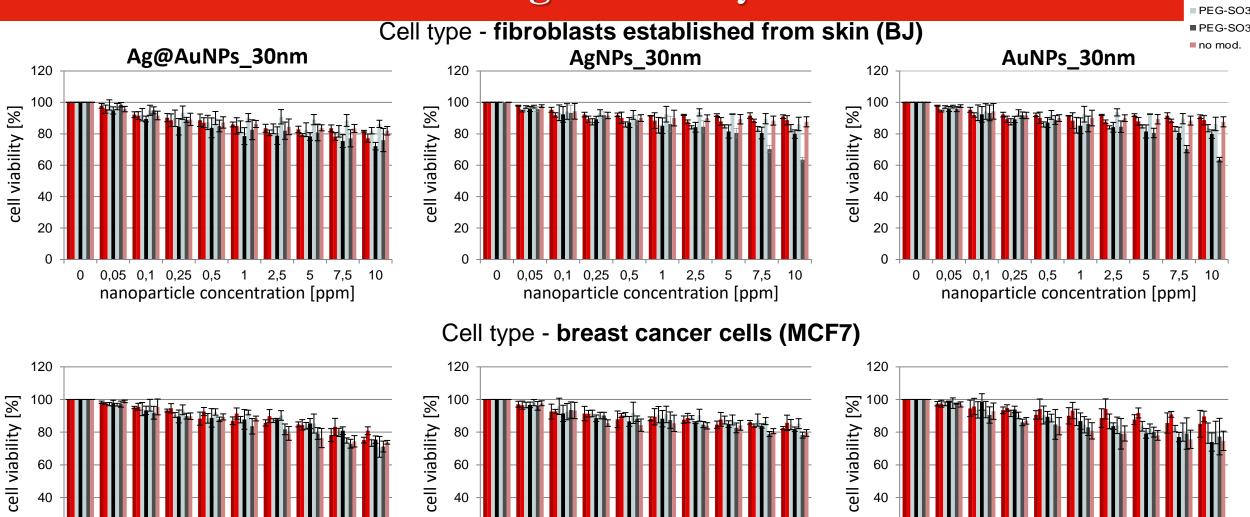
SALCULATED DEGREE OF COVERAGE [molecule/1nm²]

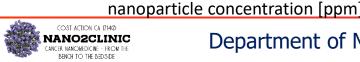
	AuNPs_30nm	Ag@AuNPs_30nm	AgNPs_30nm
MES	4,5	2,7	2,5
MUS MUS	5,5	8,9	6,5
PEG-SO ₃	**	4,5	4,5

** - at least two types of binding of molecules to the metal surface, results under analysis



Biological activity test





0,1 0,25 0,5

7,5

0,1 0,25 0,5

nanoparticle concentration [ppm]

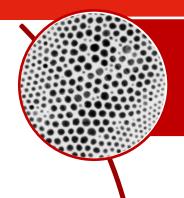
MES 5MES 20

■ MUS 5 ■ MUS 20

nanoparticle concentration [ppm]

0,1 0,25 0,5

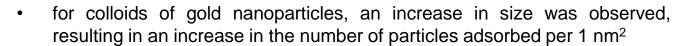
Summary



The obtaining of nanoparticles by chemical reduction method extended by the seed growth mediated synthesis leads to monodisperse colloids of metallic nanoparticles:

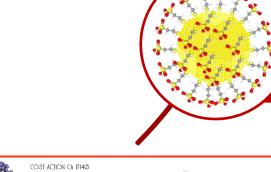
- AuNPs in size range 5 40 nm
- Ag@AuNPs in size about 30 nm
- AgNPs in size 30 nm





 the correlation between the structure of the metallic core and the amount of adsorbed molecules on the surface is difficult to describe

Knowledge of the amount of particles adsorbed on the metal surface opens the way to the conscious design of multifunctional nanoparticles



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